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SIXTH SEMESTER B.Sc. DEGREE

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(Regular/Supplementar CC15U PH6 B11 - SOLID STATE PHYSICS, SP

> Physics - Core (2015 Admission

Time: Three Hours

Section

(Answer in a word or Answer all questions. Each qu

- 1. The atomic radius for fcc-lattice is
- 2. The number of atoms present in the unit cell
- 3. The energy difference between free state of at Fermi surface.
- 4. Inside Ruby laser population inversion is ach
- 5. In He-Ne laser, lasing emission is obtained fr

Write true or false:

- 6. HCl molecule is microwave inactive.
- 7. Rotation transitions occur only in those molecules which possess a permanent electric dipole moment.
- 8. Raman scattering is due to collision between the molecules of the scatterer and the phonon.
- 9. At thermal equilibrium the coefficients of induced absorption and stimulated emission are not equal.
- 10. Soft superconductors completely expel the magnetic field from the interior of the superconducting phase.

Section B

(Answer in two or three sentences each)

- Answer *all* questions. Each question carries 2 marks.
- 11. What are Symmetry operations? Distinguish between rotation and reflection symmetry operations.
- 12. Define Glide plane and Screw axis.

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PECTROSCOPY AND LASER PHYSICS	
Course	
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	Maximum: 80 Marks
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a phrase each)	
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of hcp structure is	
electron and Coo	per pair appears as
nieved through	pumping.
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(10 x 1 = 10 Marks)

Turn Over

13. Good conductors in normal state are relatively poor superconductors. But why?

14. Distinguish between spherical top and symmetric top molecules? Give Example

15. What is meant by collision broadening?

16. Distinguish between Raman Effect and fluorescence.

17. Define population inversion and stimulated emission

(7 x 2 = 14 Marks)

Section C

(Answer in a paragraph of about half a page to one page each) Answer any *five* questions. Each question carries 4 marks.

18. Show that five-fold rotation axis does not exist in a crystal lattice.

- 19. What are Miller indices? What are the steps involved in determining Miller indices of crystal lattice.
- 20. Describe the principle underlying a SQUID. Mention its applications
- 21. Using the example of suitable molecule explain the basic process involved in the emission or absorption of microwave radiations of the electromagnetic spectrum.
- 22. Discuss vibrating diatomic molecule by considering the system as anharmonic.
- 23. Explain the isotope effect in rotational spectrum of a diatomic molecule.
- 24. Explain the principle of semiconducting laser.

(5 x 4 = 20 Marks)

Section D

(Problems: write all relevant formulas, all important steps carry separate marks) Answer any *four* questions. Each question carries 4 marks.

- 25. Obtain the Miller indices of a plane which intercepts at a,b/2,3c and 2a, ∞ , c in a simple cubic unit cell. Draw a neat diagram showing plane.
- 26. The lattice constant of a cubic lattice is 'a'. Calculate the spacing between (211), (001), (100), and (101) planes.
- 27. The critical temperature T_c for mercury with isotopic mass 199.5 is 4.185 K. Calculate its critical temperature when its isotopic mass changes to 203.4.
- 28. The first line in the rotation spectrum of carbon monoxide has a frequency of 3.8424 cm⁻¹. Calculate the rotational constant and hence the C-O bond length in carbon monoxide. Avogadro number is 6.023×10^{23}
- 29. Rotational and centrifugal distortion constants of HCl molecule are 10.593 cm⁻¹ and 5.3 X 10⁻⁴ cm⁻¹ respectively. Estimate vibrational frequency and force constant of the molecule.

- and 'R' branches of HCl.
- minimum number of chromium ions in the Ruby?

Section E

(Essays - answer in about two pages each) Answer any *two* questions. Each question carries 10 marks.

- 32. Explain Bragg's law of X-ray diffraction in crystals. How is it verified? Describe and explain X-ray spectrometer method of determining λ of X-rays.
- molecule.
- spectra of symmetric top molecule.
- 35. Explain the theory and working of He-Ne laser.

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30. The fundamental band for HCl is centered at 2886 cm⁻¹. Assuming that the inter nuclear distance is 1.276 A°. Calculate the wave number of the first two lines of each of the 'P'

31. A Ruby laser emits one J pulses of light whose wavelength is 694 nm. What is the

(4 x 4 = 16 Marks)

33. Explain rotational energy levels and rotational vibrational transitions of a diatomic

34. Briefly explain the quantum theory of Raman effect. Also discuss the rotational Raman

 $(2 \times 10 = 20 \text{ Marks})$