19P408	(Pages: 2)	Name:
		Reg. No

FOURTH SEMESTER M.Sc. DEGREE EXAMINATION, APRIL 2021 (CBCSS - PG)

CC19P CHE4 E08 - ORGANOMETALLIC CHEMISTRY

(Chemistry - Elective Course) (2019 Admission - Regular)

Time: Three Hours Maximum: 30 Weightage

Section A

Answer any *eight* questions. Each question carries 1 weightage.

- 1. Explain the ionic and covalent counting models in organometallic compounds, taking ferrocene as a representative example.
- 2. Identify **M**, if it is a first row transition element and [$\mathbf{M}(\eta^3-C_5H_5(CO)_5]$ is a stable compound.
- 3. Complete the following reactions,

(a)
$$Cp_2Co + K + 2C_2H_4 \longrightarrow ...$$
 (b) $W(CO)_6 + PPh_3^{hv} \longrightarrow ...$

- 4. What is Collman's reagent? Write one synthetic application of the reagent.
- 5. Explain the structure and aromatic behavior of (COT)²
- 6. Give one example for reductive elimination of an organometallic compound.
- 7. Draw the terminal, bridging (μ_2) , and bridging (μ_3) bonding modes of CO
- 8. Write an example for Disproportionation reaction in metal carbonyls
- 9. Write short note on oxidative addition in organometallic compounds
- 10. Write the decreasing order of back bonding and appropriate stretching vibration of the following carbonyl complexes, Mn(CO)₆⁺, V(CO)₆⁻, Cr(CO)₆; v_{CO} (cm⁻¹) 2000, 2090, 1860.

 $(8 \times 1 = 8 \text{ Weightage})$

Section B

Answer any six questions. Each question caries 2 weightage.

- 11. Discuss the features of ethylene ligand in Zeise's salt.
- 12. Discuss the bonding in the Nitrosyl (NO) metal complexes with appropriate examples.
- 13. Explain the synthesis and bonding of butadiene complexes with suitable examples.
- 14. Describe the mechanism of Wacker process.
- 15. Write short note on carbenes and carbynes complexes.

- 16. Briefly discuss the general features of migratory insertion reactions and write one example.
- 17. Explain briefly the Zeigler-Natta polymerization.
- 18. Write the Direct and Reductive synthetic methods of metal carbonyls with suitable examples.

 $(6 \times 2 = 12 \text{ Weightage})$

Section C

Answer any *two* questions. Each question carries 5 weightage.

- 19. Illustrate the bonding in ferrocene. Explain on the basis of molecular orbital energy level diagram, the cause for kinetic stability of ferrocene.
- 20. Explain the Hydrocyanation and hydrosilylation of alkenes with suitable example.
- 21. (a) With suitable diagrams, discuss the σ and π bonding in metal carbonyls.
 - (b) Explain the Monsanto acetic acid process.
- 22. (a) Explain the catalytic cycle for hydroformylation using Rhodium catalyst.
 - (b) Briefly explain the fluxional property of allyl complexes.

 $(2 \times 5 = 10 \text{ Weightage})$
