21U306

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Name:

Reg.No:

THIRD SEMESTER B.Sc. DEGREE EXAMINATION, NOVEMBER 2022

(CBCSS - UG)

(Regular/Supplementary/Improvement)

CC19U CHE3 C03 - ORGANIC CHEMISTRY

(Chemistry - Complementary Course)

(2019 Admission onwards)

Time : 2.00 Hours

Maximum : 60 Marks

Credit : 2

Part A (Short answer questions) Answer *all* questions. Each question carries 2 marks.

- 1. Explain the term heterolysis.
- 2. What is meant by sterric effect?
- 3. Which is weaker acid acetic acid or chloro acetic acid? Justify your answer.
- 4. What is meant by stereoisomerism?
- 5. Show that naphthalene is aromatic based on Huckel's rule.
- 6. Give an example of a radical halogenation reaction.
- 7. Which is more acidic-phenol or para-nitrophenol? Explain.
- 8. Explain the reaction of peopanal with HCN
- 9. Explain what happens when acetone is treated with sodium bisulphite solution.
- 10. Give the number of stereoisomers in aldotetroses.
- 11. What is fibrous protein?
- 12. Mention two uses of sandalwood oil.

(Ceiling: 20 Marks)

Part B (Short essay questions - Paragraph) Answer *all* questions. Each question carries 5 marks.

- 13. Arrange ammonia, methylamine,dimethylamine and trimethyl amine in the increasing order of their basicities. Explain the theoretical basic of your answer.
- 14. What is a racemic mixture? How does it differ from a meso form ? Explain with examples.
- 15. What is meant by nitration? Discuss the mechanism of nitration of benzene.

- 16. Explain the order of acidity of phenol, p-nitrophenol and p-methoxyphenol.
- 17. How can you convert aniline to (i)phenol, (ii) chlorobenzene.
- 18. What are epimers? Explain with examples.
- 19. Explain the source, structure and uses of citral.

(Ceiling: 30 Marks)

Part C (Essay questions)

Answer any one question. The question carries 10 marks.

- 20. Discuss the stability of different kinds of carbocations.
- 21. i) Discuss in detail, the geometrical isomerism in but-2-ene-1,4-dioic acid.ii) Discuss the methods of distinguishing geometrical isomers.

 $(1 \times 10 = 10 \text{ Marks})$
