24P213

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Name :....

Reg. No :....

SECOND SEMESTER M.Sc. DEGREE EXAMINATION, APRIL 2025

(CBCSS-PG)

(Regular/Supplementary/Improvement)

CC19P CHE2 C08 - ELECTROCHEMISTRY, SOLID STATE CHEMISTRY, AND STATISTICAL THERMODYNAMICS

(Chemistry)

(2019 Admission onwards)

Time: 3 Hours

Maximum: 30 Weightage

Section A

Answer any *eight* questions. Each question carries 1 weightage.

- 1. "The concentration term is replaced by ionic strength in the case of electrolytes" why?
- 2. Give the relationship between current density and overvoltage.
- 3. Illustrate the diagram and explain the experimental determination of overvoltage.
- 4. What is meant by crystal system?
- 5. Explain the term systematic absence in crystallography.
- 6. How does band theory distinguish semiconductors from insulators and conductors.
- 7. Explain the third law of thermodynamics using the concept of thermodynamic probability.
- 8. Derive the relation between partition functions of individual particles of a system of ideal gas and molecular partition function.
- 9. What are the assumptions of Bose- Einstein statistics?
- 10. What is thermionic emission?
- 11. State and explain debye-falkenhagen effect.
- 12. What is Sterling's approximation? What is its importance in statistical thermodynamics?

 $(8 \times 1 = 8$ Weightage)

Section **B**

Answer any *four* questions. Each question carries 3 weightage.

- 13. Relate Einstein's theory of heat capacity of solids with experimental observations.
- 14. write a note on different types of reference electrodes.
- 15. Derive an equation for the primary salt effect.
- 16. Show that fivefold axis is not possible in in crystals.

- 17. Define Hall effect. Explain how Hall effect can be used to determine the conductivity of semiconductors.
- 18. Write a note on different type of superconductors.
- 19. Derive the expession for vibrational partition function.

 $(4 \times 3 = 12 \text{ Weightage})$

Section C

Answer any two questions. Each question carries 5 weightage.

- 20. write a note on four types of fuel cells and give its advantages and limitations.
- 21. Discuss briefly the theory of polarography.
- 22. Write a brief account of the magnetic properties of solids
- 23. Explain thermodynamic probability using an example.Derive its relation to entropy. How does this relation explain the third law of thermodynamics?

 $(2 \times 5 = 10 \text{ Weightage})$
