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## FOURTH SEMESTER B.Sc. DEGREE EXAMINATION, APRIL/MAY 2015

(UG—CCSS)

Core Course—Geology

			GL 4B 07	—MIN	ERALOGY				
			(2013	Admis	ssions)				
Time :	Thre	e Hour	es and about		Maximum: 30 Weightage				
Draw neat sketches wherever necessary.									
I. Objective type questions. Answer all twelve questions.									
Choose the correct answer:									
	1 The first person to be credited with the publication of the Periodic Table :								
			Georgius Agricola.		Amedeo Avagadro.				
			Dmitri Mendeleev.	(d)	Victor Goldschmidt.				
2 Minerals that are attracted by a magnetic field are said to be:									
		(a)	Paramagnetic.	(b)	Ferromagnetic.				
		(c)	Ferrimagnetic.	(d)	Diamagnetic.				
3 The mineral which exhibits anomalous interference colours :									
		(a)	Topaz.	(b)	Garnet.				
		(c)	Zeolite.	(d)	Zoisite.				
4 Silicate structure of garnet:				salari (aratif) bara we inaini ("involati" at					
		(a)	Tecto-silicate.	(b)	Neso-silicate.				
		(c)	Phyllo-silicate.	(d)	Cyclo-silicate.				
Fill in the blanks:									
	5 Andalusite, kyanite and sillimanite and good examples of minerals showing ———.								
	<ul> <li>6 — extinction is shown by most prismatic sections of minerals crystallizing in t monoclinic and triclinic crystal systems.</li> <li>7 — is an orthorhombic amphibole.</li> </ul>								
	8	The c	hemical composition of Beryl	is —	e dinjume with the mesons on early, 18				

Name the following :-

- 9 A mineral with Van der Waals bonds.
- 10 The birefringence of Quartz.
- 11 The instrument used to determine the optic orientation, 2V and optic sign of minerals.
- 12 The mineral group to which Forsterite and Fayalite belong to.

 $(12 \times \frac{1}{4} = 3 \text{ weighta})$ 

- II. Short answer type questions. Answer all nine questions. Each question carries 1 weightage:
  - 13 Mineral and Mineraloid.
  - 14 Isomorphism.
  - 15 Opalescence and Phosphorescence.
  - 16 Refractive index.
  - 17 Polaroid.
  - 18 Anisotropic minerals.
  - 19 Sign of elongation.
  - 20 Optical properties of garnet.
  - 21 Uses of calcite and dolomite.

 $(9 \times 1 = 9 \text{ weighta})$ 

- III. Short essays type questions. Answer any five questions. Each question carries 2 weightage:
  - 22 Types of chemical bonding in minerals.
  - 23 Jolly's spring balance method.
  - 24 Parts of the Petrological microscope.
  - 25 Optical accessories and their uses.
  - 26 Uniaxial Indicatrix and Biaxial Indicatrix.
  - 27 Mineralogical composition and properties of Pyroxenes.
  - 28 Physical and Optical properties of Tourmaline.

 $(5 \times 2 = 10 \text{ weights})$ 

- IV. Essay type questions. Answer any two questions. Each question carries 4 weightage:
  - 29 Describe the structure and classification of silicates.
  - 30 Describe the mineralogy, structure, chemical composition, physical and optical propert modes of occurrence and uses of Mica.
  - 31 Give an account of the minerals Quartz and Corundum with reference to their mineralog composition, physical and optical properties, modes of occurrence and industrial uses.

 $(2 \times 4 = 8 \text{ weights})$