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## FOURTH SEMESTER B.Sc. DEGREE EXAMINATION, APRIL/MAY 2015

(U.G.—CCSS)

		Cor	e Course—C	nemistry			
		CH 4B 07—	ORGANIC	CHEMISTRY - I			
Thre	ee Hours	the bridge with		Maximum: 30 Weightage			
		Write equ	ations where	ever necessary.			
Section A							
	ltiple choice ries a weight		s type quest	ions. Answer all twelve questions. Each question			
1 The type of hybridisation of carbon in ethane is:							
	(a)	SP.	(b)	SP <sup>2</sup> .			
	(c)	SP <sup>3</sup> .	(d)	SP and SP <sup>2</sup> .			
2	2 Which of the following hydrocarbon is obtained by the Wurtz reaction between ${\rm CH_3Br}$ and ${\rm C_2H_5Br}$ with metallic sodium in ether medium?						
	(a)	Ethane.	(b)	Propane.			
	(c)	Butane.	(d)	All these.			
3 The cyclo alkane which is not expected to have ring strain is:							
	(a)	Cyclopropane.	(b)	Cyclobutane.			
	. (c)	Cyclohexane.	(d)	Cyclopentane.			
4 Which of the following compound will exhibit geometrical isomerism?							
	(a)	Butane.	(b)	2-butyne.			
	(c)	2-butene.	(d)	All these.			
5 Among the carbo cations Benzyl carbo cation (A), Allyl Carbo cation (B) and a secondary carbo cation (C), the order of the stability is:							
	(a)	A > B > C.	(b)	B > A > C.			
	(c)	C > B > A.	(d)	C>A>B.			
6	6. The molecule which exhibits ontical isomerism is:						

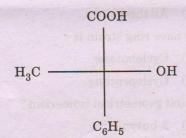
- - (a) Isobutyl chloride.
- (b) Sec. butyl chloride.
- (c) Tert. butyl chloride.
- (d) n-butyl chloride.

- 7 Calcium carbide on hydrolysis gives ———.
- 8 2-butyne on reduction with sodium in liquid ammonia gives ———
- 9 Natural rubber is a polymer of ———.
- 10 Optical isomers which are mirror images of each other are known as ———.
- 11 An example of a meta directing group is ———.
- 12 Nitration of aromatic compound is an example of substitution.

 $(12 \times \frac{1}{4} = 3 \text{ weigh})$ 

## Section B

- II. Short Answer type questions. Answer all nine questions. Each question carries a weightage
  - 13 What is resonance? Explain with an example.
  - 14 What is Kolbe reaction?
  - 15 How is polyethylene prepared?
  - 16 Draw the structure of geraniol.
  - 17 What is hydroboration?
  - 18 Draw the D and L forms of Erythrose.
  - 19 Assign the absolute configuration (R or S) of the molecule:



- 20 What is meant by C is hydroxylation?
- 21 Write any two characteristics of enantiomers.

 $(9 \times 1 = 9 \text{ weight})$ 

## Section C

- III. Short essays or paragraph questions. Answer any five questions. Each question car weightage of 2:
  - 22 What is Corey-House reaction?
  - 23 Explain Steric effect with one example.
  - 24 How will you prepare 1-butyne from acetylene?
  - 25 Discuss the E and Z designation of geometrical isomers.
  - 26 What are addition polymers? How are teflon polymers prepared?

- 27 Discuss the nitration of benzene with mechanism.
- 28 Write briefly on asymmetric synthesis.

 $(5 \times 2 = 10 \text{ weightage})$ 

## Section D

- IV. Essay questions. Answer any two questions. Each question carries a weightage of 4:
  - 29 (i) Outline Bayer's strain theory. Calculate the angle strain for various alicyclic compounds and predict their relative stability.
    - (ii) Write a note on acidity of alkynes.
  - 30 (i) What are free radicals? Give examples.
    - (ii) Write a short note on the stability of free radicals.
  - 31 (i) Define aromaticity and state Huckel's rule.
    - (ii) Discuss the structure and stability of benzene.

 $(2 \times 4 = 8 \text{ weightage})$