# FIRST SEMESTER M.Sc. DEGREE EXAMINATION, NOVEMBER 2020

(CBCSS-PG)

(Regular/Supplementary/Improvement)

## CC19P CHE1 C03 – STRUCTURE AND REACTIVITY OF ORGANIC COMPOUNDS

(Chemistry)

(2019 Admission onwards)

Time: Three Hours Maximum: 30 Weightage

#### **Section A**

Answer any *eight* questions. Each question carries 1 weightage.

- 1. Comment on the aromaticity of cyclooctatetraene, its dianion and dication.
- 2. What is meant by resonance energy?
- 3. Write a note on the importance of Marcus equation.
- 4. Differentiate between homotopic and enantiotopic hydrogens with suitable examples.
- 5. Assign the absolute configuration of following molecules.

- 6. Give the structure of the major product of the reaction between (R)-CH<sub>3</sub>-CO-CH(CH<sub>3</sub>)Ph with CH<sub>3</sub>CH<sub>2</sub>MgBr using Felkin-Ahn Model.
- 7. Draw the preferred conformations of (a) 2-chloro hexanone (b) 2-bromo-4,4-dimethyl cyclohexanone.
- 8. Compare the rate of chromic acid oxidation of *cis* and *trans* 4-t-butylcyclohexanol.
- 9. Explain why first ionization of maleic acid occurs more readily than that of fumaric acid.
- 10. Explain Bredt's rule with suitable example.

 $(8 \times 1 = 8 \text{ Weightage})$ 

### **Section B**

Answer any *six* questions. Each question carries 2 weightage.

- 11. Write a note on aromaticity of [8] and [10] annulenes.
- 12. Compare the rate etherification of isomenthol, neomenthol, isoneomenthol.
- 13. Explain Curtin-Hammett principle. Illustrate its application with suitable example.
- 14. Write a brief note on different chemical methods of resolution.

- 15. Discuss the different methods used for the determination of configuration of geometrical isomers in acyclic systems.
- 16. Explain the asymmetric hydroboration reaction using IPCBH<sub>2</sub> and IPC<sub>2</sub>BH.
- 17. Discuss the effect of conformation in the course and rate of elimination reactions with any two illustrative examples.
- 18. What are chiral auxiliaries? Illustrate the use of chiral auxiliary in asymmetric Diels-Alder reaction.

 $(6 \times 2 = 12 \text{ Weightage})$ 

#### **Section C**

Answer any two questions. Each question carries 5 weightage.

- 19. Write a note on diastereoselective aldol reaction involving *cis* and *trans* enolates. Explain the major and minor stereoisomers with the help of Zimmermann-Txaler model.
- 20. Explain why Hammett equation is considered to be a linear free energy relationship. Discuss the various parameters involved in the equation and explain their significance in the course reactions.
- 21. (a) Explain 'antiaromaticity' with suitable examples.
  - (b) What are importance rules of resonance.
- 22. (a) Give an account of isomerism of substituted biphenyls.
  - (b) Explain briefly the effect of conformation on the course and rate of  $S_N 1$  and  $S_N 2$  reactions.

 $(2 \times 5 = 10 \text{ Weightage})$ 

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