20U207	(Pages: 2)	Name:	

Reg.No:

SECOND SEMESTER B.Sc. DEGREE EXAMINATION, APRIL 2021

(CBCSS - UG)

(Regular/Supplementary/Improvement)

CC19U CHE2 B02 - THEORETICAL AND INORGANIC CHEMISTRY II

(Chemistry - Core Course)

(2019 Admission onwards)

Time: 2.00 Hours Maximum: 60 Marks

Credit: 2

Part A (Short answer questions)

Answer all questions. Each question carries 2 marks.

- 1. List the main observations from black body spectra.
- 2. Write Rydberg formula and explain the terms.
- 3. Why doesn't the wave nature of a moving cricket ball become evident to an observer?
- 4. Find $(\hat{A} \hat{B})f(x)$ if $\hat{A} = d/dx$; $\hat{B} = x^2$ and f(x) = x
- 5. State Heisenberg's Uncertainty principle.
- 6. What is meant by a well behaved wave function?
- 7. State and explain the Pauli exclusion principle.
- 8. What is Born-Oppenheimer approximation?
- 9. State variation theorem.
- 10. Write down Hamiltonian for H₂ molecule.
- 11. What is LCAO principle?
- 12. How does the MO theory explain the paramagnetism of O₂?

(Ceiling: 20 Marks)

Part B (Short essay questions - Paragraph)

Answer all questions. Each question carries 5 marks.

- 13. Starting from the expression of energy for an electron in Bohr orbit; deduce an expression for the wave number of radiations emitted for two orbits of Hydrogen atom.
- 14. Write the time dependent and time independent Schrodinger equation and explain the instances they are used.
- 15. Calculate the ground state energy of an electron confined in a one dimensional box of length 0.3 nm. Also calculate its energy when it is in n=3 level.
- 16. What are radial distribution curves? Explain the physical significance.
- 17. Calculate the bond orders of (i) B_2 (ii) B_2^+ (iii) B_2^-
- 18. Make comparison of of VB and MO theories.
- 19. What is geometry of BeH₂ molecule? How will you explain geometry on the basis of hybridization?

(Ceiling: 30 Marks)

Part C (Essay questions)

Answer any *one* questions. Each question carries 10 marks.

- 20. a) What is photoelectric effect?
 - b) Discuss how photoelectric effect established the particle nature of radiation.
 - c) The threshold wavelength of Cu is 300 nm. Calculate its threshold frequency and work function.
- 21. a) What are quantum numbers?
 - b) Discuss the significance of each quantum number.

 $(1 \times 10 = 10 \text{ Marks})$
