19U411S		(Pages 2)	Name:
		(Pages: 2)	Reg. No
	FOURTH SEMESTER	R B.Sc. DEGREE EXAM	<b>U</b>
		(CUCBCSS-UG)	,
	<b>CC15U CHE4 C04</b>	- PHYSICAL AND APP	LIED CHEMISTRY
	`	mistry - Complementary C	,
<b></b>		Admissions – Supplementa	•
Time: Three Hours			Maximum: 64 Marks
		Section A (One word)	
	Answer all o	questions. Each question c	arries 1 mark.
1.	The size of dispersed particles of a sol is in the range of		
2.	The unit of rate of a reaction is		
3.	is the most powerful chromatographic technique		
4.	Wavelength is proportional to frequency of radiation		
5.	transition is the highest energy transition in UV-Vis. spectroscopy		
6.	polymer is used in non-stick cookware		
7.	The major greenhouse gas is		
8.	Excess nitrites causes disease in human babies		
9.	The gaseous mixture above the petroleum deposits is known as		
10	). Sodium benzoate is an ex	ample of food	
			$(10 \times 1 = 10 \text{ Marks})$
		Section B (Short answer)	
	Answer any seve	en questions. Each question	n carries 2 marks.
11	. What is composting?		
12	2. What are the differences b	between hard soap and sof	t soap?
13	3. Differentiate lyophilic and	d lyophobic sols	
14	14. What is spin-spin coupling?		
15	6. What is Born–Oppenheim	ner Approximation?	
16	6. Write the important prope	erties and applications of k	Kevlar

17. What is eutrophication? What is its impact?

20. What are the health effects of hair dyes?

18. Differentiate octane number and cetane number.

19. Write short notes on antiseptics and tranquilizers

 $(7 \times 2 = 14 \text{ Marks})$ 

## **Section C** (Paragraph)

Answer any *four* questions. Each question carries 5 marks.

- 21. Define isoelectric point. Explain the stability of sols.
- 22. Derive the integrated rate equation for first order reaction.
- 23. Explain the classification of chromatographic techniques.
- 24. Define Beer–Lambert's law. Derive the equations for absorbance and Transmittance of light in the UV-Vis. Spectroscopy.
- 25. Explain the theories of color and constitution.
- 26. What is biodegradable polymer? Write short notes on any two biodegradable polymers.

 $(4 \times 5 = 20 \text{ Marks})$ 

## Section D (Essay)

Answer any two questions. Each question carries 10 marks.

- 27. Explain in detail the classification of polymers.
- 28. Explain the air pollution caused by different gases and their consequences.
- 29. Give a detailed explanation of catalysis, types and its theories.
- 30. Explain the following:
  - a) Thermal pollution and its control measures.
  - b) Radioactive pollution and its control measures.

 $(2 \times 10 = 20 \text{ Marks})$ 

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