19U610

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Name:

Reg.No:

SIXTH SEMESTER B.Sc. DEGREE EXAMINATION, APRIL 2022

(CBCSS - UG)

CC19U PHY6 B14 - MATERIALS SCIENCE

(Physics - Elective Course)

(2019 Admission - Regular)

Time: 2.00 Hours

Maximum : 60 Marks

Credit : 3

Part A (Short answer questions) Answer *all* questions. Each question carries 2 marks.

- 1. With an example explain how covalent bonding is formed?
- 2. Differentiate between crystalline and non crystalline materials.
- 3. Define atomic packing fraction
- 4. What is crystal imperfections? Explain
- 5. What is screw dislocation?
- 6. Cite two reasons why interstitial diffusion is normally more rapid than vacancy diffusion?
- 7. What is meant by abrasives?
- 8. What are Carbon nanotubes?
- 9. Differentiate between polymorphism and isomerism.
- 10. Give the relationship between interplanar spacing and miller indices.
- 11. Explain microstructure and microscopy.
- 12. Differentiate between transmission electron microscope and scanning electron microscope.

(Ceiling: 20 Marks)

Part B (Short essay questions - Paragraph) Answer *all* questions. Each question carries 5 marks.

- 13. Explain classification of materials.
- 14. Determine the linear density of [110] direction of FCC structure in terms of atomic radius.

- 15. State Fick's first law of diffusion for steady state
- The activation energy for the diffusion of carbon in chromium is 111,000 J/mol. Calculate the diffusion coefficient at 1100 K (827 ⁰C), given that D at 1400 K (1127C) is 6.25* 10 ⁻¹¹ m² /s.
- 17. What is meant by stress-strain behaviour of ceramic materials?
- 18. Explain the physical characteristics of a polymer based on its structure.
- 19. What is meant by stress relaxation and viscoelastic creep?

(Ceiling: 30 Marks)

Part C (Essay questions)

Answer any one question. The question carries 10 marks.

- 20. Explain secondary bonding with suitable examples.
- 21. Derive Bragg's law for X-ray diffraction in crystals. Describe and explain rotating cryatal method of crystal structure analysis.

 $(1 \times 10 = 10 \text{ Marks})$
