22U113

(Pages: 2)

Name:

Reg.No:

FIRST SEMESTER B.Sc. DEGREE EXAMINATION, NOVEMBER 2022

(CBCSS - UG)

(Regular/Supplementary/Improvement)

CC19U PHY1 C01 / CC20U PHY1 C01 - PROPERTIES OF MATTER AND THERMODYNAMICS

(Physics - Complementary Course)

(2019 Admission onwards)

Time: 2.00 Hours

Maximum : 60 Marks

Credit : 2

Part A (Short answer questions) Answer *all* questions. Each question carries 2 marks.

- 1. What are the theoretical limits of Poissons Ratio?
- 2. What are the drawbacks of torsion pendulum?
- 3. Define the terms neutral surface and neutral axis?
- 4. Explain the spherical shape of water surface using concept of surface energy?
- 5. How does the sign of electric charge affect the electrostatic pressure of a bubble?
- 6. Differentiate between a streamline flow and a turbulent flow of a liquid.
- 7. Explain thermal equilibrium and chemical equilibrium.
- 8. Explain adiabatic process with indicator diagram.
- 9. Explain isochoric process based on first law of thermodynamics.
- 10. Briefly explain Carnot's engine.
- 11. Explain the change in entropy during free expansion
- 12. Give Clausius Clapeyorn equation. Explain the terms.

(Ceiling: 20 Marks)

Part B (Short essay questions - Paragraph) Answer *all* questions. Each question carries 5 marks.

- 13. A spherical ball contracts in volume by 0.1 %, when subjected to a normal uniform pressure of 100 atm.Calculate the bulk modulus of the material of the ball(1 atm = 100 N/m(2))
- 14. Write a note on Brownian motion. Explain its significance.
- 15. Write a note on the viscosity of gases. Contrast it with viscosity of liquids.

- 16. Derive the expression for work done in isothermal process.
- 17. Explain Carnot engine. Give expression for efficiency of Carnot engine.
- 18. A Carnot's engine absorbs 104 calories of heat from a reservoir at 627 degree Celsius and rejects heat to a sink at 27 degree Celsius. What is its efficiency? How much work does it perform (in joule)?
- 19. Calculate the change in entropy of 5Kg water at 100 degree celsius when changes into vapour.

(Ceiling: 30 Marks)

Part C (Essay questions)

Answer any one question. Each question carries 5 marks.

- 20. How will you determine the viscosity of a liquid by Stoke's method?
- 21. What is entropy? Write its physical significance. Prove that the entropy of a system increases in an irriversible process.

 $(1 \times 10 = 10 \text{ Marks})$
