20U514	(Pages: 2)	Name:	

Reg.No:	
INCELING.	

FIFTH SEMESTER B.Sc. DEGREE EXAMINATION, NOVEMBER 2022

(CBCSS - UG)

(Regular/Supplementary/Improvement)

CC19U CHE5 B06 - INORGANIC CHEMISTRY - III

(Chemistry - Core Course)

(2019 Admission onwards)

Time: 2.00 Hours Maximum: 60 Marks

Credit: 3

Part A (Short answer questions)

Answer *all* questions. Each question carries 2 marks.

- 1. Explain the following terms with examples: (a) Pseudohalides (b) Polyhalides
- 2. What happens when XeF4 reacts with water?
- 3. What is sanitary landfil?
- 4. What are silicones?
- 5. What are phosphazenes?
- 6. Give one method of preparation of tetrasulphur tetranitride.
- 7. Explain the reaction that occurs and the nature of products obtained when AgCl is treated with Ba(NO3)2 in liquid ammonia.
- 8. What is meant by calcination?
- 9. What is the principle of behind the zone refining of metals?
- 10. Write two adverse effects caused by the pollution of water by fertilizers.
- 11. Explain the significance of determining the DO in a water sample.
- 12. Discuss the different types of light pollution.

(Ceiling: 20 Marks)

Part B (Short essay questions - Paragraph)

Answer *all* questions. Each question carries 5 marks.

13. Explain the three general methods for the preparation of interhalogen compounds.

- 14. Explain briefly about plastic pollution.
- 15. Describe how solubility product principle and common ion effect are applied in qualitative inorganic analysis.
- 16. Give an account of precipitation gravimetry.
- 17. What are intramedullary rods? Discuss their applications.
- 18. Discuss the formation and detrimental effect of photochemical smog.
- 19. Discuss the pollution of water by any two heavy metals and its adverse effects.

(Ceiling: 30 Marks)

Part C (Essay questions)

Answer any *one* question. The question carries 10 marks.

- 20. Write the main ores of aluminium. Discuss the extraction of aluminium from bauxite ore.
- 21. Explain the cause, consequences and control measures of thermal pollution.

 $(1 \times 10 = 10 \text{ Marks})$
