22U306	(Pages: 2)	Name:
		Reg.No:

## THIRD SEMESTER B.Sc. DEGREE EXAMINATION, NOVEMBER 2023

(CBCSS - UG)

(Regular/Supplementary/Improvement)

## CC19U CHE3 C03 - ORGANIC CHEMISTRY

(Chemistry - Complementary Course)

(2019 Admission onwards)

Time: 2.00 Hours Maximum: 60 Marks

Credit: 2

## Part A (Short answer questions)

Answer *all* questions. Each question carries 2 marks.

- 1. Which is more stable-but-1-ene or but-2-ene? Why?
- 2. Define hyperconjugation.
- 3. Which is the stronger acid- acetic acid or formic acid? Justify your answer.
- 4. The conformations of ethane cannot be isolated. Why?
- 5. State Huckel's rule of aromaticiy.
- 6. Explain Williamson's synthesis using a suitable example.
- 7. How is phenolphthatein prepared?
- 8. Which is more basic-ammonia or methylamine? Why?
- 9. How can you convert ethanol to acetaldehyde?
- 10. Name two polysaccharides.
- 11. Name two kinds of secondary structure normally found among proteins.
- 12. Give two uses of eucalyptus oil.

(Ceiling: 20 Marks)

Part B (Short essay questions - Paragraph)

Answer *all* questions. Each question carries 5 marks.

- 13. Arrange ammonia, methylamine,dimethylamine and trimethyl amine in the increasing order of their basicities. Explain the theoretical basic of your answer.
- 14. Describe conformational isomerism with regard to ethane.
- 15. Explain the term directive influence of substituent groups in aromatic electrophilic substitution reactions.

- 16. Explain the order of acidity of ethyl alcohol, isopropylalcohol and tert-butylalcohol.
- 17. How can you prepare aniline from nitrobenzene? How can you convert aniline to phenyl isocyanide?
- 18. Discuss the secondary structure of protein.
- 19. Explain the source, structure and uses of menthol.

(Ceiling: 30 Marks)

## Part C (Essay questions)

Answer any *one* question. The question carries 10 marks.

- 20. Discuss the aspects regarding the mechaism, kinetics and factors affecting the reactivity of of SN<sup>1</sup> reactions?
- 21. Explain any five synthetic applications of benzenediazoniumchloride.

 $(1 \times 10 = 10 \text{ Marks})$ 

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