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FIRST SEMESTER M.Sc. DEGREE EXAMINATION, NOVEMBER 2016

(Regular/Supplementary/Improvement)

(CUCSS-PG)

CC15P CH1 C02 – ELEMENTARY INORGANIC CHEMISTRY

(Chemistry)

(2015 Admission Onwards)

Time: Three Hours

Maximum: 36 Weightage

Section A

(Answer all questions. Each question carries a weightage of one)

- 1. Give a method of preparation of Borazines
- 2. What happens 1, 2-dicarboclosodecaborane (12) is heated?
- 3. Write a note on diagonal relationship.
- 4. To which class of compounds, polyhedron skeleton electron pair theory is applicable? Explain
- 5. Give one example for photo nuclear reactions.
- 6. What are Phosphazenes.
- 7. State and explain Octet rule.
- 8. Write notes on formal charges.
- 9. What is the name of correlation diagram represents energy of molecular orbitals with change in bond angle?
- 10. Explain why oxygen and fluorine do not participate in hypervalency. ?
- 11. Explain super heavy elements with examples
- 12. What is Zeolite?

(**12** x **1** = **12** weightage)

Section B

(Answer any eight questions. Each question carries a weightage of two)

- 13. What are the differences between 4f and 5f orbitals?
- 14. Write a short note on Hyper valence
- 15. Write notes on Pourbaix diagram
- 16. Explain different types of carbides with suitable examples
- 17. What is Bent rule of hybridization? Explain.
- 18. Discuss synthesis, properties and applications of Silicones.
- 19. Discuss structure of S_2N_2 and S_4N_4
- 20. Write briefly on nuclear liquid drop model.
- 21. Explain closo, nido and arachno boranes.

- 22. Discuss the important chemical reactions that occur in liquid ammonia and FSO₃H
- 23. Give the structure and preparation of B_5H_9
- 24. Write a note on superacids.

(8 x 2 = 16 weightage)

Section C

(Answer any 2 questions. Each question carries a weightage of four)

- 25. Explain the classification of Lewis acids into hard and soft acid and bases. Comment on the chemical consequences of this concept in the formation of coordination compounds
- 26. Discuss structure and bonding in Phosphazenes.
- 27. Write notes on Heteropoly, Isopoly acids of W and Mo and its applications
- 28. Explain (a) leveling effect of solvents (b) Discuss Frost diagram.

(2 x 4 = 8 weightage)