FIRST SEMESTER M.Sc. DEGREE EXAMINATION, NOVEMBER 2019 (CUCSS PG) CC19P CHE1 C01 – QUANTUM MECHANICS AND COMPUTATIONAL CHEMISTRY (Chemistry)

(2019 Admission Regular)

Time: Three Hours

Maximum: 30 Weightage

Section A

Answer any *eight* questions. Each question carries 1 weightage.

- 1. Show that $\Psi = sin(k_1x) \cdot sin(k_2y) \cdot sin(k_2)$ is an eigenfunction of ∇^2 . What is the eigenvalue?
- 2. Give two experiments each showing the particle nature of waves and wave nature of particles.
- 3. Show that $[\hat{x}^n, \hat{p}_x] = -\frac{h}{2\pi} n x^{n-1}$
- 4. The state of a system is described by a time dependent wave function $\Psi(x, t)$ but the average value of any physical quantity is independent of time. Why?
- 5. An electron is confined in a one-dimensional box of length 1Å. Calculate its ground state energy. Is quantization of energy levels observable? (mass of electron = 9×10^{-31} kg).
- 6. What are even and odd functions? Illustrate by taking wave functions of one dimensional harmonic oscillator.
- 7. Prove that the nonexistence of zero point energy in planar rigid rotator is not in violation of Heisenberg's uncertainty principle.
- 8. What are Legendre polynomials and where they are used?
- 9. Justify the choice of spherical polar coordinates for solving Hydrogen like atom.
- 10. Write the Slater determinant for Li atom.

(8 x 1 = 8 Weightage)

Section B

Answer any six questions. Each question carries 2 weightage.

- 11. An electron in a box of width L undergoes a transition from the lowest energy level (n=1) to the first excited level (n=2). The wavelength of light absorbed in this transition was determined to be 650 nm. Calculate the width of the box.
- 12. Draw rough graphs of ψ and ψ^2 for the v = 4 state of the one dimensional harmonic oscillator and explain tunneling effect.
- 13. Discuss the space quantization of angular momentum.

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- 14. Find the eigenvalues and eigenfunctions of a particle on a ring.
- Show that the orbitals 2p_x, 2p_y, and 2p_z have their maximum and minimum values along X, Y, and Z axes respectively. Locate their nodal planes.
- 16. Apply perturbation method to ground state of the helium atom.
- 17. Write a note on post HF methods.
- 18. Construct Z matrix for methanol.

(6 x 2 = 12 Weightage)

Section C

Answer any *two* questions. Each question carries 5 weightage.

- 19. Setup the Schrödinger equation for hydrogen atom and separate the variables to obtain R, Θ and Φ equations. Write the general solution of these equations.
- 20. What do you mean by first order perturbation methods? Determine the ground state energy of a particle in one dimensional box with slanted bottom using perturbation method.
- 21. (a) Explain in detail various steps involved in the Hatree-Fock method for atoms.(b) Write the STO of Fe atom.
- 22. Set up the Schrodinger equation for particle on a sphere and solve the equation to find out the wave function for the case of m=0. Also get the expression for the energy.

(2 x 5 = 10 Weightage)
