

**FIRST SEMESTER M.Sc. DEGREE EXAMINATION, NOVEMBER 2025**

(CBCSS - PG)

(Regular/Supplementary/Improvement)

**CC19PCHE1C02 - ELEMENTARY INORGANIC CHEMISTRY**

(Chemistry)

(2019 Admission onwards)

Time : 3 Hours

Maximum : 30 Weightage

**Section A**Answer any ***eight*** questions. Each question carries 1 weightage.

1. Arrange  $\text{BF}_3$ ,  $\text{BI}_3$ ,  $\text{BBr}_3$  and  $\text{BCl}_3$  in the increasing order of acid strength. Justify your answer.
2. What are super acids? Explain using an example.
3. What are carboranes?
4. Write notes on diagonal relationship.
5. What is the action of borazine on  $\text{HCl}$ ?
6. What are styx numbers?
7. Distinguish isopoly anion from heteropoly anions.
8. What is a Frost diagram? What information do we get from this diagram?
9. Explain the merits and demerits of Liquid drop model.
10. Differentiate photonuclear reactions from thermonuclear reactions.
11. What is lithography? Give two examples for photoresists used in lithographic applications.
12. How is graphene and rGO related?

**(8 × 1 = 8 Weightage)****Section B**Answer any ***four*** questions. Each question carries 3 weightage.

13. Explain the Drago-Wayland equation and the E, C parameters.
14. Derive the Styx code for  $\text{B}_2\text{H}_6$ .
15. How is  $\text{S}_2\text{N}_2$  prepared? Discuss its structural characteristics.

16. Applying wade's rules classify the following boranes by structural type. a)  $\text{B}_4\text{H}_{10}$  b)  $\text{B}_4\text{H}_8$
17. What are super heavy elements? Give two examples? Give the method of production of any two super heavy elements.
18. Write briefly on Scintillation detectors.
19. Write a short note on surface plasmon resonance (SPR).

**(4 × 3 = 12 Weightage)**

### **Section C**

Answer any ***two*** questions. Each question carries 5 weightage.

20. Discuss briefly the reactions which take place in non- aqueous sulphuric acid,  $\text{SO}_2$ ,  $\text{N}_2\text{O}_4$ .
21. Discuss the structure and bonding in Diborane. How it is synthesized? Explain its reaction with ammonia.
22. What is an Ellingham diagram? Discuss briefly about its applications and limitations.
23. What are the commonly employed top-down and botto-up approaches deployed in the synthesis of nanomaterials?

**(2 × 5 = 10 Weightage)**

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